City of Porterville CONSUMER CONFIDENCE REPORT

For period 1/1/11 to 12/31/11

Este informe contiene informacion muy importante sobre su agua potable de beber. Traduze este informacion or si tiene preguntas, pueden hablar con Bert Yarbrough a 782-7518.

In order to ensure that tap water is safe to drink, USEPA and the State Department of Health Services (Department) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. Department regulations also establish limits for contaminants in bottled water that provide the same protection for public health. The City of Porterville tests its water at each well site and at numerous locations throughout the distribution system on a regular basis to guarantee compliance with all state and federal standards.

City of Porterville water comes from 34 municipal water wells located throughout the city. Before being pumped into the distribution system, a disinfectant is added to the water to protect you from potential microbial contaminants.

An assessment of all the drinking water sources for the City of Porterville was completed in November of 2011. Of the 34 wells, six have been determined to be vulnerable to PCE contamination, and seven have been determined to be vulnerable to nitrate contamination, and one vulnerable to DBCP contamination; however, none of these constituents have been detected at concentrations over the drinking water standards. A copy of the complete assessment is available at the City Corporation yard, 555 N. Prospect Street.

We are required to monitor your drinking water for specific contaminants on a regular basis. Results of regular monitoring are an indicator of whether or not our drinking water meets health standards. For additional water quality data, please contact Jeff Friedman at 782-7518. Your concerns can also be addressed to the Porterville City Council. Meetings are held at 6:30 p.m. on the first and third Tuesdays of each month at City Hall, 291 N. Main Street. Council sessions are open to the public. Property owners, with any type of tenants, please make copies and distribute them to your tenants and/or post on your community board if available.

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

Contaminants that may be present in source water include: microbial contaminants such as viruses and bacteria; inorganic contaminants such as salts and metals; pesticides and herbicides that may come from a variety of sources; organic chemical contaminants, including synthetic and volatile organic chemicals, that are by-products of industrial processes and petroleum production, and radioactive contaminants that can be naturally occurring.

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the USEPA's Safe Drinking Water Hotline (800) 426-4791.

IMPORTANT REMINDER FOR AQUARIUM OWNERS AND HOME DIALYSIS PATIENTS

To meet USEPA regulations the water supply will contain chlorine. Residents with aquariums or fish ponds should remove the chlorine with water conditioning chemicals or granular activated carbon. Contact your local tropical fish store to determine the best water treatment for your fish. If you are receiving kidney dialysis treatment, please contact your doctor or dialysis technician to be sure that the equipment is adequately removing the chlorine.

OTHER PRECAUTIONS THE PUBLIC SHOULD CONSIDER

Some people may be more vulnerable to contaminants in drinking water than the general population. Immune-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. USEPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline (800) 426-4791.

Definitions:

Public Health Goal (PHG) - The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

Maximum Contaminant Level Goal (MCLG) - The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency (USEPA).

Maximum Contaminant Level (MCL) - The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

Action Level (AL) - The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

Primary Drinking Water Standard - Primary MCLs, specific treatment techniques adopted in lieu of primary MCLs, and monitoring and reporting requirements for MCLs that are specific in regulations. For more Definitions see bottom of page 2.

Table 1: Primary Standards and Unregulated Contaminants												
Constituent		MCL		PHG (MCLG)	Range of Detections	Averag	e '	Year Sampled		Typical Source of Contaminant		
Volatile Organic Contamina	nts											
1,1-Dichloroethylene (DCE) (ppb)		6		10	0-0.8	0.04	201	2011		Discharge from industrial chemical factories		
Tetrachloroethylene (PCE) (ppb)		5		.06	0-3.4	0.7	201	2011		Discharge from factories, dry cleaners, and auto shops (metal degreaser)		
Synthetic Organic Contamii	nants incl	uding Pe	sticides	, Insectici	des, and Herb	oicides			l .			
Dibromochloropropane (DBCP) (ppt)		200		1.7	0-30	2.4	201	2011		Runoff/leaching from former use of nematocide on soybeans, cotton, vineyards, tomatoes, and tree fruit.		
Heptachlor (ppb)		10		8	0-10 1			2011		Residue of banned insecticide.		
Inorganic Contaminants	1				<u> -</u>	1 -0-		1100.000				
Aluminum (ppm)		1		0.6	0-0.440	0.013	201	2011		Erosion of natural deposits		
		10		0.004	0005	0.0002		2011		Erosion of natural deposits		
Arsenic (ppb)								2011		<u> </u>		
Barium (ppm)		2.0		2	0-0.240	0.124		-		Erosion of natural deposits		
Fluoride (ppm)	Fluoride (ppm)			1.0	0-0.1	.0054	201	1		Erosion of natural deposits		
Nitrate (ppm)		45		45	0-36.8	17.6 2		1			ning from fertilizer use; ptic tanks and sewage; erosion its	
NITRATE: Infants below the age of six months who drink water containing nitrate in excess of the MCL may quickly become seriously ill and, if untreated, may die because high nitrate levels can interfere with the capacity of the infant's blood to carry oxygen. Symptoms include shortness of breath and blueness of the skin. High nitrate levels may also affect the oxygen-carrying ability of the blood of pregnant women.												
Radionuclides												
Gross Alpha (pCi/L)	15			0	3.1-11	6.5 2010 & 2011		Erosion of natural deposits				
Jranium (pCi/L)		20		0.43	0-9.25	4.7	.7 2010 & 2011		Erosion of natural deposits		al deposits	
Unregulated Contaminants NL												
Boron (ppb)		n/a		1000	70-140	92	201	2011		Erosion of natural deposits		
Disinfection Byproducts, [Disinfecta	nt Resid	duals an	d Disinfect	ion Precursors	s			<u> </u>			
Total Trihalomethanes(TTHM) (ppb)		80	80 n/a		2.8-6.9	8-6.9 4.85		1	Byprodu	Byproduct of drinking water chlorination		
Chlorine (ppm)		4.0	4.0 4		0.28 - 0.42	0.42 0.36		1	Disinfed	Disinfectant added for treatment		
Table 2: Micro Biological Conto		minant	ninants									
				MCLG or	r Sampling							
Constituent	ent MC		<u>.</u>		PHG	Frequency		Amount Detected		Possible Source		
Total Coliform Bacteria	Presence i		e in less than 5% of samples		0	15 samples per week		0.33% of s in 2011.	of samples taken l.		Naturally present in the environment	
	ms, the sar	nple is th	en retest	ed. The sam	ole is retaken up-	-stream, do	wn-stream	and at the origi	inal sample locat	tion to cor	present. If a representative system firm and/or identify initial result. If s are absent.	
Table 3: Lead and Copp	er (201	0)	1						_			
Canadian	# of sar			ercentile	· ·	# of sites exceeding AL		44.01.0		Densible Commo		
Constituent Lead (ppb)		collected		level detected		exceeding AL 0		MCLG 2		Possible Source Internal corrosion of household water plumbing systems; erosion of natural deposits		
Copper (ppm)	38		0.17		0		1.3	0.3	Internal con		rosion of household water plumbing	
LEAD: If present, elevated levels of lead can cause serious health problems, especially for pregnant woman and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. The City of Porterville is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.												
Table 4: Secondary Sto												
Constituent		MC			Range of detections			Average			Year Sampled	
Chloride (Cl) (ppm)		500	500		7.9 -27			14.2			2011	
Color (units)		15			1-5			1,2			2011	
Sodium (ppm)			n/a		14 - 32			24			2011	
					251 - 521			<u> </u>				
Specific Conductance (micromhos)			1600					384			2011	
Sulfate (SO4) (ppm)			500		6.5 - 22			12.5			2011	
Total Dissolved Solids (TDS) (ppm)		100	1000		120 - 305			214		2011		

100 - 210

0.1 - 4.1

Total Hardness as (CaCO3) (ppm)

Turbidity (Lab) (units)

n/a

5

150

0.37

2011

2011